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Total synthesis of the marine cytotoxic caulibugulones A-D

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Abstract—We report the first total synthesis of the cytotoxic marine alkaloids caulibugulone A–D. This synthesis confirmed the assigned structures and provided sufficient material for further biological testing.

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Marine organisms such as sponges, tunicates, coelenterates, and phytoplankton have proven to be a valuable source of biologically active secondary metabolites.¹ Very recently, Milanowski et al.² identified four new isoquinoline quinone derivatives and two iminoquinones, termed caulibugulones A–F (Fig. 1) from the marine bryozoan *Caulibugula intermis* Harmer (Bugulidae) and reported interesting cytotoxic activity. A number of isoquinoline quinones, including the renierones^{3,4} and cribrostatins^{5,6} have been isolated and reported to exhibit antimicrobial and antitumor activity. Quinoneimines are well represented in marine alkaloids, such as isobatzellines,⁷ makaluvamines,^{8,9} and secobatzellines A–B,¹⁰ and exhibit interesting cytotoxic and antimicrobial profiles. The structural characteristic of caulibugulones is an isoquinoline-5,8dione carrying a substituted amino group in position-7 and substituted at position-6 by hydrogen, bromine or



Figure 1. ${}^{a}IC_{50}$ are expressed in $\mu g/mL$ against the murine $IC-2^{WT}$ cell line in an in vitro antiproliferative assay.²

Keywords: Caulibugulones A-D; Isoquinoline-5,8-diones; Cytotoxicity.

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Scheme 1. Reagents and conditions: (a) sulfanilic acid, NaNO₂, H_2SO_4 , AcOH, AcONa; (b) NaOH, Na₂S₂O₄; (c) $K_2Cr_2O_7$, H_2SO_4 ;¹¹ (d) NH₂CH₂CH₂OH, CeCl₃, EtOH, rt; (e) MeNH₂, CeCl₃, EtOH, rt; (f) NBS, MeOH, rt; (g) NCS, MeOH, rt.





chlorine (caulibugulones A–D, 1–4). Caulibugulones E and F are analogues of caulibugulone A (1) carrying an imine group at position-5. As part of an ongoing effort on synthesis of new bioactive heterocycles, the present paper describes the first de novo synthesis of caulibugulones A–D.

The key intermediate 5,8-isoquinolinedione (7) was prepared in three steps (30% overall yield) according to the methodology described by Joseph and Joullie ^{11,12} starting from the commercially available 5-aminoisoquinoline (5) as outlined in Scheme 1. Compound 7 was converted regioselectively to 1^{13} (74%) and 4^{13} (50%) by oxidative amination with methylamine or 2-aminoethanol in ethanol in the presence of CeCl₃ (Scheme 1). The regioselectivity of this reaction can be explained by resonance stabilization of the 1,4-adduct at C-7 via the pyrinoid nitrogen (Fig. 2), which favors substitution at C-7. Subsequent treatment of 1 with *N*-bromosuccinimide or *N*-chlorosuccinimide in methanol provided the desired compound 2^{13} and 3^{13} in 97% and 94% yield, respectively (Scheme 1).

In conclusion, the synthesis of the naturally occurring caulibugulones A–D confirmed the assigned structures and provided sufficient material for further biological testing.

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- 13. Spectral data of selected compounds: The ¹H and ¹³C spectrum for caulibugulones A–D (1–4) were identical to those described.² Compound 1: purification (SiO₂, CH₂Cl₂/MeOH, 95:5), red solid, mp = 208–210 °C (dec), HRMS calcd for $C_{10}H_8N_2O_2$ (M+): 188.0585, found

188.0587. Compound **2**: purification (SiO₂, CH₂Cl₂/ MeOH 95/5), red solid, mp = 183–185 °C (dec), HRMS calcd for C₁₀H₇BrN₂O₂ (M+): 265.9690, found 265.9694. Compound **3**: purification (SiO₂, CH₂Cl₂/MeOH, 95:5), red solid, mp = 151–153 °C (dec), HRMS calcd for C₁₀H₇ClN₂O₂ (M+): 222.0196, found 222.0198. Compound **4**: purification (SiO₂, CH₂Cl₂/MeOH, 90:10), red solid, mp = 173–175 °C, HRMS calcd for C₁₁H₁₀N₂O₃ (M+): 218.0691, found 218.0684.